Problems on Ionic and Net Ionic Equations

For each of the following sets of reactants;

- a. Write the balanced chemical equation including states of the reactants and products assuming the reaction takes place in water
- b. Write the ionic form of the equation
- c. Write the net ionic equation
- 1. NaF + Ca(NO₂)₂ \rightarrow

2.
$$Sr(C_2H_3O_2)_2 + Al_2(SO_4)_3 \rightarrow$$

3.
$$(NH_4)_3PO_4 + FeSO_4 \rightarrow$$