

## Chemical Nomenclature Answers to Sample Test

- |   |  |   |  |
|---|--|---|--|
| 1. Potassium Sulfate                              |  | 2. Calcium Chloride                           |  |
| 3. Iron(II) Oxide                                 |  | 4. Carbon(IV) Oxide                           |  |
| 5. Lead(II) hypobromite                           |  | 6. Hydrogen Chloride                          |  |
| 7. Copper(II) nitrate                             |  | 8. Manganese(II) Bicarbonate                  |  |
| 9. Mercury(II) Phosphate                          |  | 10. Potassium dihydrogen Phosphate            |  |
| 11. Zinc Dichromate                               |  | 12. Silver hydroxide                          |  |
| 13. Ammonium nitrite                              |  | 14. Chlorine(VII) Oxide                       |  |
| 15. Barium perchlorate                            |  | 16. Carbon(IV) chloride                       |  |
| 17. Aluminum cyanide                              |  | 18. Magnesium sulfate heptahydrate            |  |
| 19. Cuprous chloride                              |  | 20. Nickelic Thiosulfate                      |  |
| 21. Carbon monoxide                               |  | 22. Dibromine Pentoxide                       |  |
| 23. Hydrosulfuric Acid                            |  | 24. Hydroiodic Acid                           |  |
| 25. Sulfurous Acid                                |  | 26. Hypoiodous acid                           |  |
| 27. Pernitric Acid                                |  | 28. Perchloric Acid                           |  |
| 29. Phosphoric Acid                               |  | 30. Acetic Acid                               |  |
|   |  |   |  |
| 31. $\text{Na}_3\text{N}$                         | 32. $\text{Cl}_2\text{O}_5$                | 77. $\text{Co}(\text{NO}_2)_2$                | 78. $\text{H}_2\text{CO}_3$                              |
| 33. $\text{AlAs}$                                 | 34. $\text{SO}_2$                          | 79. $(\text{NH}_4)_2\text{SO}_3$              | 80. $\text{Ni}_2(\text{CO}_3)_3$                         |
| 35. $\text{H}_2\text{SO}_3$                       | 36. $\text{PCl}_5$                         | 81. $\text{Ca}(\text{ClO})_2$                 | 82. $\text{HNO}_3$                                       |
| 37. $\text{Mn}(\text{C}_2\text{H}_3\text{O}_2)_2$ | 38. $\text{H}_3\text{PO}_4$                | 83. $\text{Cu}(\text{BrO})_2$                 | 84. $\text{Cr}(\text{CN})_3$                             |
| 39. $\text{BaF}_2$                                | 40. $\text{H}_3\text{P}$                   | 85. $\text{H}_3\text{PO}_2$                   | 86. $\text{HNO}_2$                                       |
| 41. $\text{FeI}_2$                                | 42. $\text{Cu}_2\text{O}$                  | 87. $\text{CsHSO}_3$                          | 88. $\text{SiBr}_2$                                      |
| 43. $\text{Fe}(\text{HSO}_4)_2$                   | 44. $\text{Br}_2\text{O}_5$                | 89. $\text{HBrO}_4$                           | 90. $\text{AgHSO}_4$                                     |
| 45. $\text{CoCl}_3$                               | 46. $\text{FeCl}_3$                        | 91. $\text{SO}_3$                             | 92. $\text{HClO}$  |
| 47. $\text{HIO}$                                  | 48. $\text{P}_2\text{O}_5$                 | 93. $\text{HBr}$                              | 94. $\text{Rb}_2\text{S}$                                |
| 49. $\text{HgSiO}_3$                              | 50. $\text{CCl}_4$                         | 95. $\text{Zn}(\text{H}_2\text{PO}_4)_2$      | 96. $\text{Fe}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ |
| 51. $\text{LiOH}$                                 | 52. $\text{HNO}_4$                         | 97. $\text{HBr}$                              | 98. $\text{HClO}_4$                                      |
| 53. $\text{Sr}(\text{BrO}_3)_2$                   | 54. $\text{OF}_6$                          | 99. $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ | 100. $\text{CuBr}$                                       |
| 55. $\text{Cu}(\text{ClO}_3)_2$                   | 56. $\text{Ag}_2\text{C}_2\text{O}_4$      |   |  |
| 57. $\text{HClO}_2$                               | 58. $\text{Al}_2(\text{Cr}_2\text{O}_7)_3$ |   |  |
| 59. $\text{K}_2\text{CrO}_4$                      | 60. $\text{N}_2\text{O}_4$                 |   |  |
| 61. $\text{Sn}(\text{OH})_2$                      | 62. $\text{CO}$                            |   |  |
| 63. $\text{Fe}(\text{MnO}_4)_3$                   | 64. $\text{H}_2\text{Te}$                  |   |  |
| 65. $\text{Na}_2\text{O}_2$                       | 66. $\text{Zn}(\text{NO}_3)_2$             |   |  |
| 67. $\text{H}_2\text{S}$                          | 68. $\text{NaHCO}_3$                       |   |  |
| 69. $\text{BaS}_2\text{O}_3$                      | 70. $\text{CsCl}$                          |   |  |
| 71. $\text{Cu}(\text{BrO}_4)_2$                   | 72. $\text{HClO}_3$                        |   |  |
| 73. $\text{NaClO}_4$                              | 74. $\text{H}_2\text{SO}_4$                |   |  |
| 75. $\text{KBrO}_2$                               | 76. $\text{HC}_2\text{H}_3\text{O}_2$      |   |  |